



Cambridge IGCSE™

COMBINED SCIENCE**0653/43**

Paper 4 Theory (Extended)

May/June 2024**MARK SCHEME**Maximum Mark: 80

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge International will not enter into discussions about these mark schemes.

Cambridge International is publishing the mark schemes for the May/June 2024 series for most Cambridge IGCSE, Cambridge International A and AS Level and Cambridge Pre-U components, and some Cambridge O Level components.

This document consists of **10** printed pages.

Generic Marking Principles

These general marking principles must be applied by all examiners when marking candidate answers. They should be applied alongside the specific content of the mark scheme or generic level descriptions for a question. Each question paper and mark scheme will also comply with these marking principles.

GENERIC MARKING PRINCIPLE 1:

Marks must be awarded in line with:

- the specific content of the mark scheme or the generic level descriptors for the question
- the specific skills defined in the mark scheme or in the generic level descriptors for the question
- the standard of response required by a candidate as exemplified by the standardisation scripts.

GENERIC MARKING PRINCIPLE 2:

Marks awarded are always **whole marks** (not half marks, or other fractions).

GENERIC MARKING PRINCIPLE 3:

Marks must be awarded **positively**:

- marks are awarded for correct/valid answers, as defined in the mark scheme. However, credit is given for valid answers which go beyond the scope of the syllabus and mark scheme, referring to your Team Leader as appropriate
- marks are awarded when candidates clearly demonstrate what they know and can do
- marks are not deducted for errors
- marks are not deducted for omissions
- answers should only be judged on the quality of spelling, punctuation and grammar when these features are specifically assessed by the question as indicated by the mark scheme. The meaning, however, should be unambiguous.

GENERIC MARKING PRINCIPLE 4:

Rules must be applied consistently, e.g. in situations where candidates have not followed instructions or in the application of generic level descriptors.

GENERIC MARKING PRINCIPLE 5:

Marks should be awarded using the full range of marks defined in the mark scheme for the question (however; the use of the full mark range may be limited according to the quality of the candidate responses seen).

GENERIC MARKING PRINCIPLE 6:

Marks awarded are based solely on the requirements as defined in the mark scheme. Marks should not be awarded with grade thresholds or grade descriptors in mind.

Science-Specific Marking Principles

1 Examiners should consider the context and scientific use of any keywords when awarding marks. Although keywords may be present, marks should not be awarded if the keywords are used incorrectly.

2 The examiner should not choose between contradictory statements given in the same question part, and credit should not be awarded for any correct statement that is contradicted within the same question part. Wrong science that is irrelevant to the question should be ignored.

3 Although spellings do not have to be correct, spellings of syllabus terms must allow for clear and unambiguous separation from other syllabus terms with which they may be confused (e.g. ethane / ethene, glucagon / glycogen, refraction / reflection).

4 The error carried forward (ecf) principle should be applied, where appropriate. If an incorrect answer is subsequently used in a scientifically correct way, the candidate should be awarded these subsequent marking points. Further guidance will be included in the mark scheme where necessary and any exceptions to this general principle will be noted.

5 'List rule' guidance

For questions that require ***n*** responses (e.g. State **two** reasons ...):

- The response should be read as continuous prose, even when numbered answer spaces are provided.
- Any response marked *ignore* in the mark scheme should not count towards ***n***.
- Incorrect responses should not be awarded credit but will still count towards ***n***.
- Read the entire response to check for any responses that contradict those that would otherwise be credited. Credit should **not** be awarded for any responses that are contradicted within the rest of the response. Where two responses contradict one another, this should be treated as a single incorrect response.
- Non-contradictory responses after the first ***n*** responses may be ignored even if they include incorrect science.

6 Calculation specific guidance

Correct answers to calculations should be given full credit even if there is no working or incorrect working, **unless** the question states 'show your working'.

For questions in which the number of significant figures required is not stated, credit should be awarded for correct answers when rounded by the examiner to the number of significant figures given in the mark scheme. This may not apply to measured values.

For answers given in standard form (e.g. $a \times 10^n$) in which the convention of restricting the value of the coefficient (a) to a value between 1 and 10 is not followed, credit may still be awarded if the answer can be converted to the answer given in the mark scheme.

Unless a separate mark is given for a unit, a missing or incorrect unit will normally mean that the final calculation mark is not awarded. Exceptions to this general principle will be noted in the mark scheme.

7 Guidance for chemical equations

Multiples / fractions of coefficients used in chemical equations are acceptable unless stated otherwise in the mark scheme.

State symbols given in an equation should be ignored unless asked for in the question or stated otherwise in the mark scheme.

Question	Answer			Marks												
1(a)(i)		<table><tr><th>label</th><th>name</th><th>function</th></tr><tr><td>A</td><td>(sepal)</td><td>to protect / to support (the flower / bud / ovary)</td></tr><tr><td>(C)</td><td>anther</td><td>(produces) pollen / male gamete</td></tr><tr><td>D</td><td>stigma</td><td>(site of pollination)</td></tr></table>	label	name	function	A	(sepal)	to protect / to support (the flower / bud / ovary)	(C)	anther	(produces) pollen / male gamete	D	stigma	(site of pollination)		3
	label	name	function													
	A	(sepal)	to protect / to support (the flower / bud / ovary)													
	(C)	anther	(produces) pollen / male gamete													
	D	stigma	(site of pollination)													
	...															
	...															
	...															
	...															
	one mark for each correct row															
1(a)(ii)	any one from: (large) petals ; anther inside (petals / flower) ; stigma inside (petals / flower) ;			1												
1(b)(i)	any two from: flagellum ; enzymes ; AVP ;			2												
1(b)(ii)	(nucleus of sperm cell) fuses / joins ; with <u>nucleus</u> of, ovum / egg ;			2												

Question	Answer	Marks
2(a)(i)	activation energy correctly labelled on both diagrams ; energy change of reaction shown by upward arrow for first diagram ; energy change of reaction shown by downward arrow for second diagram ;	3

Question	Answer	Marks										
2(a)(ii)	<table><tr><td>true</td><td>false</td></tr><tr><td></td><td>✓</td></tr><tr><td>✓</td><td></td></tr><tr><td>✓</td><td></td></tr><tr><td></td><td>✓</td></tr></table> All correct = 2 marks 2 to 3 correct = 1 mark	true	false		✓	✓		✓			✓	2
true	false											
	✓											
✓												
✓												
	✓											
2(a)(iii)	2NO + O ₂ → 2NO ₂ ;	1										
2(b)	weak forces between molecules / weak intermolecular forces ; need low amounts of energy to change from liquid to gas / low amount of energy to, separate molecules / overcome / break the force ;	2										

Question	Answer								Marks								
3(a)(i)	<table><tr><td>gamma radiation</td><td></td><td>ultraviolet</td><td></td><td></td><td></td><td>microwaves ;</td><td>radio waves</td></tr></table>								gamma radiation		ultraviolet				microwaves ;	radio waves	1
gamma radiation		ultraviolet				microwaves ;	radio waves										
3(a)(ii)	sunburn / skin cancer / skin damage / eye damage ;								1								
3(b)(i)	speed = distance ÷ time <i>in any form</i> OR $37\,000 \div 3.0 \times 10^5$; $12(.3) \times 10^{-2}$ OR 0.12 (s) ;								2								

Question	Answer	Marks
3(b)(ii)	conversion km to m seen ; $v = f\lambda$ in any form / $3 \times 10^8 \div 12 \times 10^9$; 0.025 (m) ;	3

Question	Answer	Marks
4(a)(i)	pancreas labelled ;	1
4(a)(ii)	amylase breaks down or digests, starch / carbohydrate, or amylase has an active site complementary to starch ; protein (shape), is not complementary to / will not fit <u>active site</u> ;	2
4(a)(iii)	amino acids ;	1
4(b)(i)	lack of sunlight ;	1
4(b)(ii)	soft, bones / teeth ;	1
4(b)(iii)	thin walls / permeable walls / short diffusion distance ; to allow exchange of material / to allow substances to, diffuse / pass through the walls (between cells and blood) ;	2

Question	Answer	Marks
5(a)(i)	zinc sulfate ; zinc oxide / zinc hydroxide ;	2
5(a)(ii)	hydrogen ; water ;	2
5(b)	carbon dioxide ; water ;	2

Question	Answer	Marks
5(c)	full range / universal, indicator ; compare (colours) with chart / colour corresponds to a number ;	2
5(d)	add, calcium carbonate / limestone / calcium hydroxide / (slaked) lime / add a base ;	1
6(a)(i)	arrow pointing to right, touching bus and same length as arrow Q ;	1
6(a)(ii)	friction / air resistance ;	1
6(b)	force P = 75000 N or weight of bus = 75000 N ; no vertical motion / arrow R is same length as arrow P / no resultant force / R and P are balanced forces ;	2
6(c)(i)	acceleration = initial slope of graph OR 12 / 100 ; 0.12 ; m / s ² ;	3
6(c)(ii)	distance while accelerating = $\frac{1}{2} 100 \times 12$ OR 600 (m) ; distance at constant speed = 150×12 OR 1800 (m) ; total distance = (1800 + 600) = 2400 (m) ;	3
6(c)(iii)	(distance = 2650 – 2400 =) 250 (m) ;	1

Question	Answer	Marks
7(a)(i)	both / they, transport (substances) ; and either xylem / X , also provides support / transports mineral ions / transports up the stem ; or phloem / Y , transports sugars / amino acids / transports up and down the stem ;	2

Question	Answer	Marks
7(a)(ii)	by evaporation (from the surface) ; water vapour moves to / out of, stomata ; (water vapour passes through stomata) by diffusion ;	3
7(b)	tip ; elongation ; gravity and light ;	3
7(c)(i)	small fish and clam ;	1
7(c)(ii)	tertiary carnivore large ;; <i>all correct = 2 marks</i> <i>two correct = 1 mark</i>	2

Question	Answer	Marks
8(a)(i)	they all have similar chemical properties / all contain a (carbon-carbon) double bond ;	1
8(a)(ii)	alkenes: C_nH_{2n} ; alkanes: C_nH_{2n+2} ;	2
8(b)	six carbon atoms bonded in a row (bonded with hydrogen atoms) ; 6 C and no double bonds and 12 H correctly bonded and continuation lines shown ;	2

Question	Answer	Marks
8(c)(i)	4 C atoms and 10 H atoms bonded in a row ; correct butane structure drawn to show single bonds ;	2
8(c)(ii)	any two from: high temperature ; high pressure ; catalyst ;	2

Question	Answer	Marks
9(a)	volts ; electrons / charge ;	2
9(b)(i)	power = $V \times I$ or 2.4×0.50 ; 1.2 (W) ;	2
9(b)(ii)	$Q = It$ or $9000 = 0.5 \times t$ or $(t =) 9000 \div 0.5$; (time = $9000 \div 0.5$) = 18000 s ; seconds to hours = 5(.0) h ;	3
9(c)(i)	lamps in parallel (instead of in series) ;	1
9(c)(ii)	greater (<i>no mark</i>) (reason) less resistance or current from source / total current, is sum of (separate) currents in, each lamp / each branch ;	1